

Polymer Communication

# Synthesis and characterization of polystyrene-substituted [70]fullerene

Yu Chen\*, Jinxia Wang, Yanghui Lin, Ruifang Cai, Zu-En Huang

*Department of Chemistry, Fudan University, Shanghai 200433, People's Republic of China*

Received 14 December 1998; received in revised form 10 May 1999; accepted 7 June 1999

## Abstract

Polystyrene-substituted [70]fullerene was synthesized via the reaction of  $C_{70}$  with living polystyrene anions in the THF– $C_6H_6$  mixed solvents. The covalent graftings of the polystyrene arms onto the  $C_{70}$  core result in the enhancement of the absorption degree at longer wavelengths, and in the decrease of the thermal stability and the glass transition temperature of the parent polymer. The other characterization results are also described. © 1999 Elsevier Science Ltd. All rights reserved.

*Keywords:*  $C_{70}$ ; Polystyrene; Polystyrene-substituted  $C_{70}$

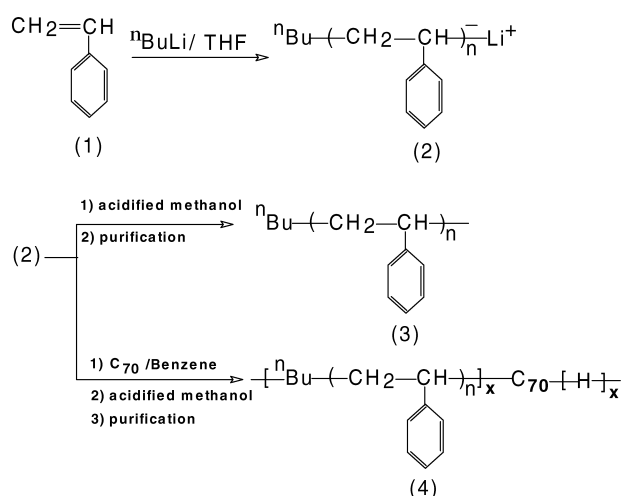
## 1. Introduction

Over the past seven years investigations into the covalent [60]fullerene chemistry have made enormous progress and led to a rich variety of tailor-made building blocks for many technological applications [1–4]. In comparison, much less work has been done on [70]fullerene chemistry due to its high cost, lower symmetry and lower abundance. Unlike  $C_{60}$ , the prolate spheroidal fullerene  $D_{5h}$ - $C_{70}$  [5], which consists of 12 pentagonal faces fused to 25 hexagonal faces, has a lower electron-withdrawing character than  $C_{60}$  [6], and exhibits a striking variety of structural features with five types of non-equivalent carbon atoms, eight distinct types of carbon–carbon bonds, two types of five-membered rings, and three types of six-membered rings in which the bond distance alternation is marked in type I rings, moderate in type II rings, and non-existent in the equatorial type III rings [7]. Theoretical and experimental methods show that  $C_{70}$  is a closed shell system with a HOMO–LUMO gap of 1.76 eV (to be compared to 1.68 eV in  $C_{60}$ ) which accounts for the stability of this cluster [8,9]. Its lower symmetry relaxes the forbiddenness of some absorption and emission processes [10], resulting in the visible absorption of  $C_{70}$  being markedly stronger than that of  $C_{60}$ , and giving rise to a more complicated pattern of isomeric structures than for  $C_{60}$  upon functionalization of the fullerene core [11]. The chemistry of  $C_{70}$  is thus more complicated than that of  $C_{60}$  due to differences in reactivity between the various double bonds. This difference can be estimated by considering  $\pi$

bond orders or the local spheroid curvature, which is expressed as pyramidalization angles of the corresponding carbon atoms and is greatest at the poles [12]. Like  $C_{60}$ ,  $C_{70}$  has also been functionalized in many ways and its possible organic reactions mainly include: (1) hydrogenation [13]; (2) oxidation reactions [14,15]; (3) halogenations [16,17]; (4) organometallic reactions [7,18,19]; (5) cycloadditions [20–26]; (6) the addition of nucleophiles [27,28]; and (7) the addition of radicals [29]. Noteworthy is the general tendency for additions at 6–6 bonds in areas of highest local curvature near the poles of  $C_{70}$ , and for bis-additions at 6–6 bonds in apposite hemispheres [12]. Compared to  $C_{60}$ ,  $C_{70}$  shows a higher selectivity in multiple nucleophilic additions and a somewhat lower affinity for radicals. In recent years, there are already some papers concerning the functionalization of  $C_{70}$ , however, only a few of them reported the polymeric modification of [70]fullerene. Cao and Webber [30] described simply a method for incorporation of  $C_{70}$  into polystyrene (PS) by direct free radical copolymerization under routine conditions. Although a great deal remains to be done to characterize fullerenes as comonomers in free radical polymerization, this method may be of interest to a wide range of researchers working in the area of fullerene chemistry. Cai and coworkers reported the synthesis of photoconductive  $C_{70}$  chemically modified poly(*N*-vinylcarbazole) via the reaction of  $C_{70}$  with the carbanion intermediates of polymer [31]. This novel polymer is highly soluble in some common organic solvents, and has significantly different spectroscopic properties from both poly(*N*-vinylcarbazole) and  $C_{70}$ . Here, we report the covalent attachment of “living” PS carbanions prepolymerized onto the  $C_{70}$  core and describe the effect of

\* Corresponding author.

*E-mail address:* chym@fudan.edu.cn (Yu Chen)



$C_{70}$  chemical modification on the physical and chemical properties of PS.

## 2. Experimental

All the operations for synthesis prior to the termination reaction were carried out under purified dry argon by use of standard *schlenk* tubes. Into a stirring solution of styrene (6 ml) in THF (80 ml) at 0°C under argon was added 0.31 M *n*-butyllithium (4 ml) via a syringe. The color of the solution changed rapidly from colorless to red-orange which is characteristic of styryl carbanions in THF. After stirring for 40 min, the “living” PS solution was divided into two equal parts. The first part of solution was directly terminated by acidified methanol. The molecular weight average ( $M_w$ ) and molecular number average ( $M_n$ ) of pure PS obtained are  $7.7 \times 10^3$ ,  $4.5 \times 10^3$ , respectively. Then, a diluted solution of  $C_{70}$  (100 mg, 0.119 mmol, purity >99%) in dry, degassed benzene (100 ml) was added dropwise to the another part of solution within 1 h, followed by stirring for 1 week. The resultant polymer was precipitated with acidified methanol, redissolved in THF, filtered to remove any insoluble impurity (if any) and reprecipitated with methanol to give a brownish yellow solid (this procedure was repeated twice). Solvent residues were removed by freeze drying under vacuum for several hours. Gel permeation chromatography of this product gave  $M_n = 2.49 \times 10^4$  and  $M_w/M_n = 1.3$ . The synthetic route of polystyrene-substituted [70]fullerene is shown in Scheme 1.

## 3. Results and discussion

Recently, studies on the preparation and structural characterization of styrene-based polymeric  $C_{60}$  derivatives have increased rapidly. The results show that the fullerene molecule may be incorporated covalently into the PS backbone by use of [32–37]: (1) radical-initiated polymerization

reactions; (2) sodium naphthalene-initiated anionic copolymerization reactions; (3) cycloaddition reaction of azido-substituted polystyrenes with fullerene; (4) Friedel–Crafts type reactions; (5) amine addition reactions; and (6) reaction of polymeric living carbanions with fullerene. Of all polymerization methods, the addition of living anionic polymers pre-generated onto the  $C_{60}$  core could offer a unique opportunity to control both the number of chains grafted onto the fullerene and the molar mass and polymolecularity of the grafts [37]. The maximum number of grafted chain depends on the reactivity of the carbanion borne by the chain. Although the chemistry of  $C_{70}$  is more complicated than that of  $C_{60}$ , their mode of addition reaction are very similar and differ only in their rates of reaction with certain species. The living PS carbanions thus also could add to the carbon–carbon bonds of [70]fullerene. This idea is confirmed by our experimental results. By assuming that the *n*-butyllithium added is completely used to initiate anionic polymerization of styrene monomer and in the addition reaction  $C_{70}$  is also completely consumed, the degree of polymerization and average molecular weight of pure PS obtained may be roughly calculated as 42 and 4377, respectively. The average number of grafted chains and average molecular weight of polystyrene-substituted [70]fullerene (i.e.  $C_{70}$ -PS copolymer) are about 5 and  $2.36 \times 10^4$ , respectively. These calculated values are comparatively in agreement with the experimental values.

The UV–VIS absorption spectra of  $C_{70}$ , PS and  $C_{70}$ -PS copolymer indicated that the main absorption bands of  $C_{70}$  in *n*-hexane ( $C_{70}$  is not soluble in any polar solvents) are located at 210, 231, 245, 312, 327, 354, 373 and 468 nm, respectively. Unlike  $C_{60}$ , the visible absorption of  $C_{70}$  is markedly stronger than that of  $C_{60}$ , probably because its lower symmetry relaxes the forbiddenness of the longest wavelength transition [10]. For polystyrene-substituted [70]fullerene, in the UV-region, the strong absorption bands at 242, 248, 254, 262 and 268 nm corresponding to the copolymer are in general agreement with those of pure PS, but their relatively peak intensities are different. The covalent graftings of multiple PS chains onto the  $C_{70}$  core apparently result in the enhancement of the absorption degree at longer wavelengths. Further, it is found that, polystyrene-substituted [70]fullerene polymer loses specific bands of  $C_{70}$ . Similar to the UV–VIS spectrum exhibited by  $C_{60}$ -styrene copolymer prepared in sodium naphthalene-initiated anionic polymerization reactions [32], the somewhat structured absorption band of free  $C_{70}$  is also replaced by a steadily decreasing curve, typical for substituted  $C_{70}$ .

$^{13}\text{C}$  NMR results of the  $C_{70}$ -PS copolymer and pure PS obtained in deuterated chloroform showed that, in the pure *n*-butyl-terminated PS, the resonant signals of aliphatic carbons locate at  $\delta = 40.6, 42.8, 43.6, 44.0, 14.0,$  and  $22.5$  ppm, respectively. Among these peaks, the final two peaks are attributed to the *n*-butyl-terminated groups, the peak at 40.6 ppm may be ascribable to  $\alpha$ -C, the other three peaks correspond to the triad splitting  $\beta$ -C. The six

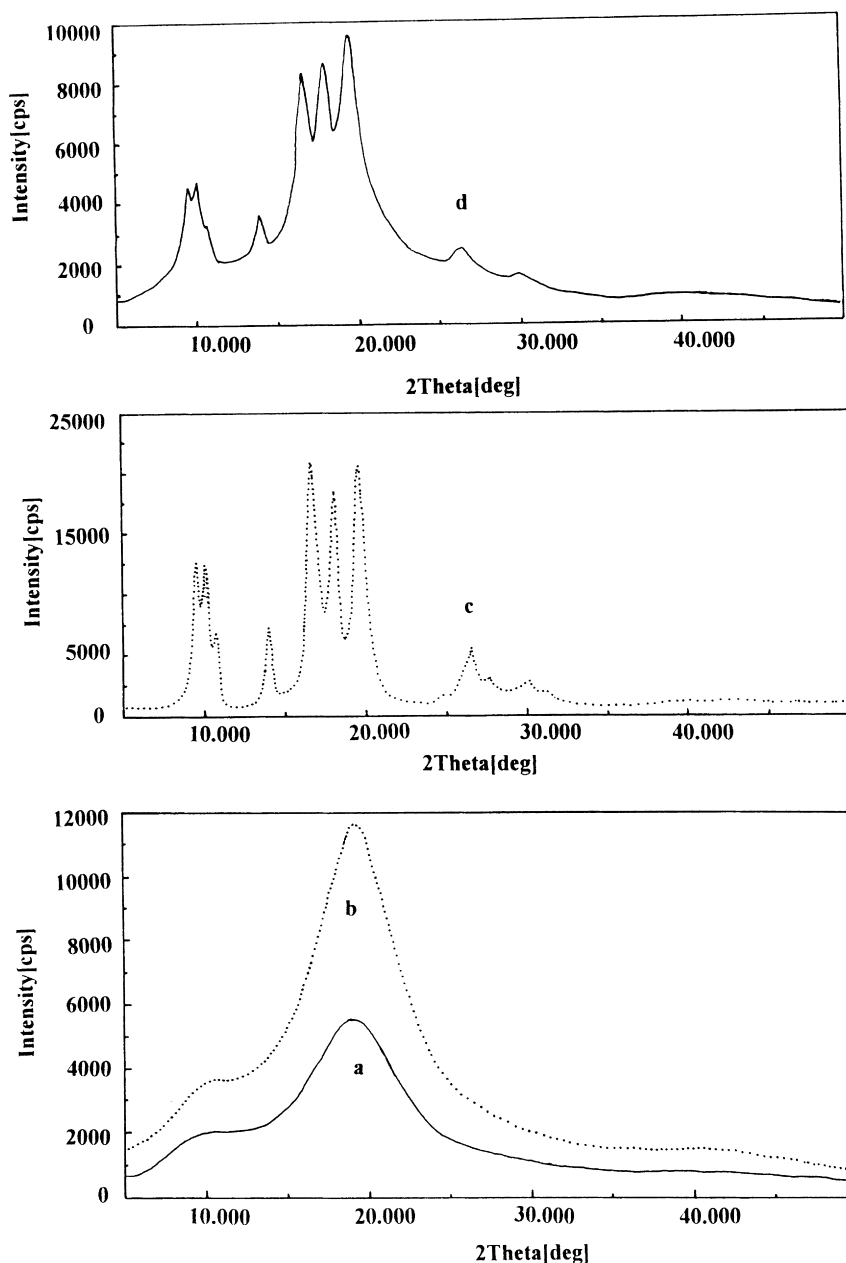


Fig. 1. X-ray powder diffraction diagrams from: (a) PS; (b) C<sub>70</sub>-PS copolymer; (c) C<sub>70</sub>; and (d) C<sub>70</sub>-doped PS.

carbon signals at 125.4, 126.0, 127.7, 129.2, 145.7, and 145.3 ppm in the aromatic region may be assigned to C-4 (125.4, 125.9 ppm), C-3, C-5 (127.7 ppm), C-2, C-6 (129.2 ppm) and C-1 (145.7, 145.3 ppm) on phenyl rings of PS, respectively. For the copolymer, the observed <sup>13</sup>C NMR spectrum is essentially the same as that of pure PS and gave a little structural information due to the relatively lower content in the copolymer of [70]fullerene with large relaxation time. In the aliphatic region of <sup>13</sup>C NMR spectrum of copolymer, four new peaks located at 25.6, 29.7, 30.2, and 68.0 ppm were detected. The chemical shift at  $\delta = 68.0$  ppm is attributed to the sp<sup>3</sup> hybridized carbons of substituted C<sub>70</sub> cages [26]. Due to the effect of stronger

electron affinity of C<sub>70</sub> on the aliphatic carbons of PS backbone, the chemical shifts of partial  $\alpha$ -C and  $\beta$ -C on the adjacent PS chain segments attached directly to the C<sub>70</sub> cores shift to the up-field considerably, and appear at 25.6 ppm for  $\alpha$ -C, 29.7 and 30.2 ppm for  $\beta$ -C, respectively. Similarly, because of the overwhelming contributions of multiple PS chains, which may lead to a great encapsulation to C<sub>70</sub> core, in the polymer structure, the structure changes can hardly be observed by the IR spectrum.

Differential scanning calorimetric measurements of the glass transition temperatures of polymers show that like PS, the C<sub>60</sub>-PS copolymer has also a single glass transition temperature (53.9°C as against 83.4°C for PS), implying a

single-phase system. However, covalent graftings of living PS arms onto the C<sub>70</sub> core result in the decrease of the glass transition temperature of the parent polymer. Perhaps a better rationalization might run as follows. In contrast to [60]fullerene with high symmetry, additions at 6–6 bonds of olivary [70]fullerene often occurred in areas of highest local curvature near the poles, while further multiple additions at 6–6 bonds occurred in apposite hemispheres [12]. In this case, the relatively lower steric hindrance effect from inter-chains of polymer in copolymer leads to the increase of the flexibility of polymer chains, and consequently results in the decrease of the glass transition temperature of the parent polymer. Likewise, the thermal stability of the parent polymer is also decreased due to the chemical modification with C<sub>70</sub>, and distinct stages of weight loss on the pure PS and PS substituted [70]fullerene: 320–480°C and 200–550°C, respectively, are observed.

The X-ray powder diffraction scan for the C<sub>60</sub>-PS copolymer is shown compared with those from pure PS and C<sub>70</sub>-doped PS in Fig. 1. The eight relatively stronger sharp diffraction peaks of C<sub>70</sub> centered at  $2\theta$  16.74° ( $I/I_0 = 100$ ), 19.68° ( $I/I_0 = 99$ ), 18.12° ( $I/I_0 = 87$ ), 9.62° ( $I/I_0 = 60$ ), 10.18° ( $I/I_0 = 60$ ), 14.04° ( $I/I_0 = 35$ ), 10.82° ( $I/I_0 = 32$ ) and 26.54° ( $I/I_0 = 26$ ). The XRD pattern from C<sub>70</sub>-PS copolymer was very similar to that of pure PS. After multiple PS arms were attached onto the C<sub>70</sub> core, two main diffraction peaks at  $2\theta = 10.10^\circ$  and  $19.04^\circ$  still keep unchanged, indicating that the periodicity of the main arrangement of atoms or atomic groups in the copolymer remains unchanged on the incorporation of C<sub>70</sub> cage into the polymer. However, it is seen that the diffracted intensities of the peaks at  $2\theta = 10.10^\circ$  and  $19.04^\circ$ , especially at  $19.04^\circ$   $2\theta$ , were increased and consequently the peak form narrowed. The XRD spectrum of C<sub>70</sub>-doped PS seems to be only a simple superposition of two spectra corresponding to C<sub>70</sub> and pure PS, further indicating that in copolymer C<sub>70</sub> the molecule has indeed incorporated covalently into the PS backbone.

In summary, we have demonstrated the synthesis of polystyrene-substituted [70]fullerene in the heterogeneous medium and described the initial results of the structural characterization. Further characterization and study of the properties of this material are currently being investigated.

### Acknowledgements

This research was supported by the National Natural Science Foundation of China (Nos. 29671007, 29890216), the Young Teacher Foundation of Fudan University and the State Education Commission Doctoral Foundation of China. We would particularly like to express our thanks to Dr Yi-Min Hou for her generous help.

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